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Cite as: J. Appl. Phys. **115**, 17B530 (2014); <https://doi.org/10.1063/1.4869158>

Submitted: 21 September 2013 . Accepted: 17 January 2014 . Published Online: 20 March 2014

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## Effect of ball milling and dynamic compaction on magnetic properties of Al<sub>2</sub>O<sub>3</sub>/Co(P) composite particles

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(Presented 7 November 2013; received 21 September 2013; accepted 17 January 2014; published online 20 March 2014)

The evolution of the magnetic properties of composite Al<sub>2</sub>O<sub>3</sub>/Co(P) particles during ball milling and dynamic compaction is investigated. To prepare starting composite particles, the Al<sub>2</sub>O<sub>3</sub> granules were coated with a Co<sub>95</sub>P<sub>5</sub> shell by electroless plating. The magnetic and structural properties of the composite particles are characterized by scanning electron microscopy, X-ray diffraction, and the use of the Physical Property Measurement System. The use of composite core-shell particles as starting powder for mechanoactivation allows to decrease treatment duration to 1 h and to produce a more homogeneous bulk sample than in the case of the mixture of Co and Al<sub>2</sub>O<sub>3</sub> powders. The magnetic properties of the milled composite particles are correlated with changes in the microstructure. Reduction in grain size of Co during milling leads to an increase of the volume fraction of superparamagnetic particles and to a decrease of the saturation magnetization. The local magnetic anisotropy field depends on the amount of hcp-Co phase in sample. The anisotropy field value decreases from 8.4 kOe to 3.8 kOe with an increase in milling duration up to 75 min. The regimes of dynamic compaction were selected so that the magnetic characteristics—saturation magnetization and coercive field—remained unchanged. © 2014 AIP Publishing LLC.

[<http://dx.doi.org/10.1063/1.4869158>]

### I. INTRODUCTION

Ceramometals (cermets), consisting of uniformly dispersed ferromagnetic metallic particles into an oxide matrix, have been the subject of intensive study due to their unique combination of magnetic and mechanical properties: enhanced microwave absorption,<sup>1</sup> giant magnetoresistance,<sup>2</sup> and good catalytic properties<sup>3</sup> without losing of mechanical properties of monolithic Al<sub>2</sub>O<sub>3</sub> ceramic. Usually cermets are manufactured from a mixture of metallic and oxide powders, that is, pressed and sintered at high temperatures;<sup>4</sup> several chemical methods are used as well.<sup>5,6</sup> However, retaining nanometric grain sizes during powder consolidation techniques is difficult. Thus, producing bulk nanostructured materials still remains relevant. Another way to prepare cermets is based on the mechanical alloying of various metals with aluminum or Al<sub>2</sub>O<sub>3</sub> powder.<sup>7,8</sup> The ball milling has been demonstrated to be able to perform mechanochemical synthesis (MS), to grind powders to nanometric scale, and to produce large quantities of material for a low cost. It was found<sup>9,10</sup> that the desired mechanical properties (fracture toughness) of bulk cermets can be achieved by introducing Co particles with certain hcp/fcc phase ratio. However, to prepare cermet by traditional MS, long treating time is needed, reaching up to 16–20 h of treating in a high energy ball mill. In previous research,<sup>11</sup> we show that the use of composite core-shell particles as starting powder for MS decreases treatment duration

to 1–2 h. In this paper, we utilize the same approach to produce the Al<sub>2</sub>O<sub>3</sub>/Co(P) cermet (i.e., we use the core-shell particles as starting powder for MS in order to reduce the time needed to prepare the composite particles with optimal hcp/fcc phase ratio).

In the present work, we report on the evolution of structural and magnetic properties of Al<sub>2</sub>O<sub>3</sub>/Co(P) composite particles during ball milling and dynamic compaction processes. We show that the magnetic properties of the composites depend strongly on the microstructure created during the mechanical activation in the planetary ball mill, and that the use of composite core-shell particles allows us to produce a more homogeneous bulk sample than in the case of the mixture of Co and Al<sub>2</sub>O<sub>3</sub> powders.

### II. EXPERIMENTAL

The initial Al<sub>2</sub>O<sub>3</sub>/Co(P) composite particles consist of granular Al<sub>2</sub>O<sub>3</sub> cores (50 mass %), surrounded by a shell of Co<sub>95</sub>P<sub>5</sub> particles. The Co-P particles was deposited on Al<sub>2</sub>O<sub>3</sub> surface by electroless plating as described in our previous study.<sup>12</sup> The milling of the composite particles was performed in planetary ball mill AGO-2U with stainless-steel vials and balls during 15, 30, 45, 60, 75, and 90 min. The ball-to-powder weight ratio is 20:1. The Co<sub>95</sub>P<sub>5</sub> particles were also milled under the same conditions. The morphology and the composition of the powders were analyzed using scanning electron microscopes (SEM) (Carl Zeiss EVO 60 and Hitachi TM3000 with Xflash 430 H detector, Bruker)

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and an energy-dispersive spectrometer (EDS) (INCA, Oxford Instruments). The crystalline structure of the composite powders was determined using a DRON-4 X-ray diffractometer operating with Cu  $K\alpha$  radiation. The field and low-temperature dependences of magnetization,  $M(T,H)$ , were measured in the external field range up to 50 kOe and at 4.2–300 K using a Physical Property Measurement System (PPMS). The bulk composite samples were prepared by dynamic compaction.<sup>13</sup>

### III. RESULTS AND DISCUSSION

The SEM images of the initial composite particles are presented at Fig. 1(a). It can be seen that the shape of the Co(P) particles is nearly spherical for  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  composite powders. The average  $\text{Co}_{95}\text{P}_5$  particle size supported on the surface of  $\text{Al}_2\text{O}_3$  granules is 200 nm. Figures 1(b) and 1(c) show SEM images of the  $\text{Al}_2\text{O}_3/\text{Co(P)}$  composite particles ball milled during 45 and 75 min, respectively. After milling the particles were plastically deformed, and flattened particles were broken to smaller pieces. The majority of the particles contained both  $\text{Al}_2\text{O}_3$  and Co(P), as revealed by EDS analysis (Fig. 2). After 45 min of milling, the average particle size increases, while we observe a wider particle size distribution. It should be noted that ductility of  $\text{Al}_2\text{O}_3/\text{Co(P)}$  composite increases after treating for 45 min. When milling time reaches 75 min the volume fraction of particles with size smaller than 100 nm increases.

The XRD patterns of the  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  powders after ball milling for various treatment durations are shown in Fig. 2(d). The initial  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}$  powder XRD pattern corresponds to the hcp Co phase and  $\text{Al}_2\text{O}_3$ . For the sample with 15 min milling time, XRD peaks corresponding to the Co hcp phase and those corresponding to the Co fcc phase are presented together. As the milling time reaches 45 min, the XRD analysis indicates the presence of Co fcc phase, while the Co hcp peaks further weaken. Such transformations of Co hcp phase to Co fcc phase during milling have been frequently observed.<sup>14,15</sup> The XRD analysis reveals that the pure  $\text{Co}_{95}\text{P}_5$  powder milling causes following transformations: hcp Co(P)  $\rightarrow$  fcc Co(P)  $\rightarrow$  amorphous state  $\rightarrow$  hcp Co(P) after 30, 60, and 90 min of milling duration, respectively. Evidently, the hard  $\text{Al}_2\text{O}_3$  particles prevent the amorphous state formation during ball milling of composite particles.

The hysteresis loops of the initial  $\text{Al}_2\text{O}_3/\text{Co(P)}$  composite powders and milled powders are shown in Fig. 3. The coercivity values insignificantly grow after ball milling

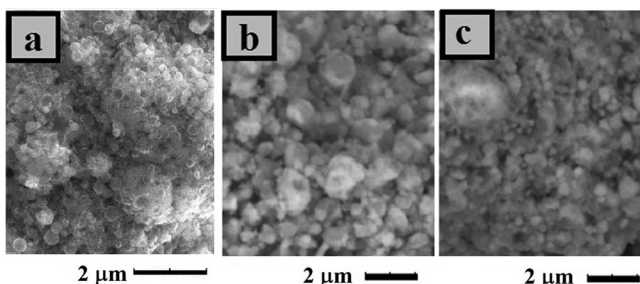


FIG. 1. Scanning electron micrograph of the  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  composite particles before (a), and after ball milling: (b) during 45 min, (c) 75 min.

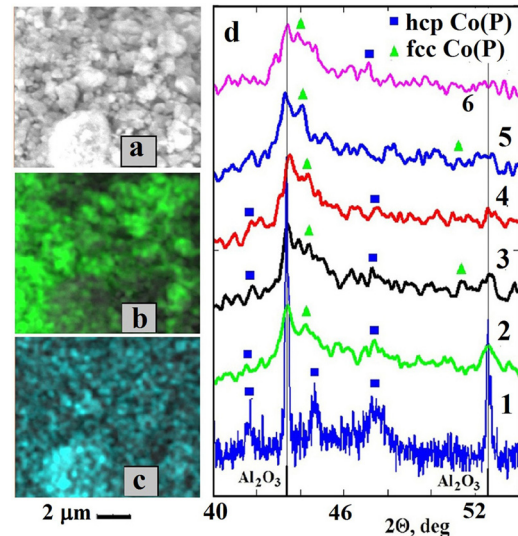


FIG. 2. SEM image of  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  particles milled for 45 min. (a) The images in (b) and (c) are the corresponding Co and Al EDX mappings of the composite particles milled for 45 min; (d) XRD spectra of the  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  composite powders for (1) initial and (2) 15 min, (3) 30 min, (4) 45 min, (5) 60 min, and (6) 75 min of milling time.

( $H_c \sim 300$  Oe for 75 min milled powder). The inset on Fig. 3 shows the temperature dependence of magnetization for the  $\text{Al}_2\text{O}_3/\text{Co(P)}$  composites before and after treating in the planetary ball mill. We fit the  $M(T)$  curve with the following equation:

$$M(T) = A(1 - BT^{3/2}) + CL\left(\frac{D}{T}\right), \quad (1)$$

where  $M(T)$  is magnetization of  $\text{Al}_2\text{O}_3/\text{Co(P)}$  particles at temperature  $T$ ,  $B$  is the Bloch constant,  $L(x)$  is the Langevin function,  $A = M_f V_f$ ,  $C = M_{sp} V_{sp}$ , and  $D$  are fitting constants. Here,  $M_f$  is the average magnetization of the ferromagnetic phase,  $V_f$  is the volume fraction of this phase,  $M_{sp}$  is the average magnetization of the superparamagnetic phase,  $V_{sp}$  is the volume fraction of this phase. The first term corresponds to

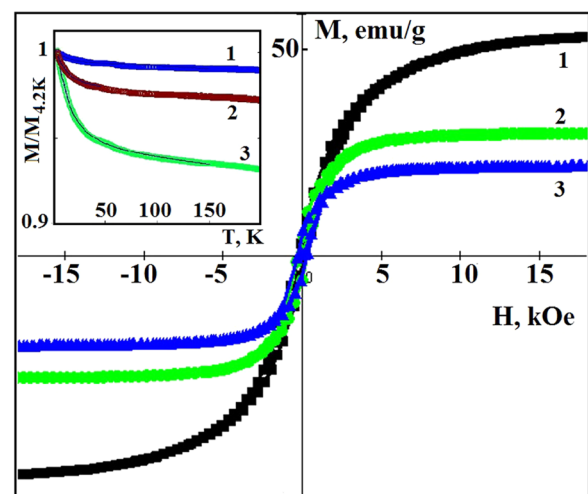


FIG. 3. The hysteresis loops measured at room temperature for  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  composite particles (1-black rhomb), and ball milled for 15 min (2-green circle), for 75 min (3-blue triangle). Inset show the temperature dependence of magnetization for the  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  (1) composites and powders ball milled during 15 min (2) and 75 min (3).

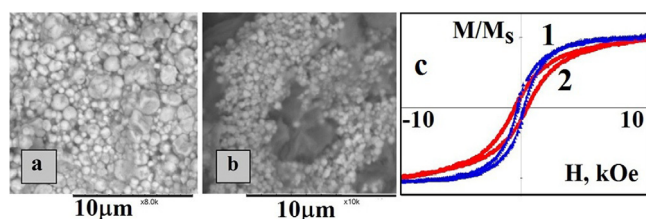


FIG. 4. SEM images of the bulk materials produced by dynamic compaction of the  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  composite particles (a) and mixture of  $\text{Al}_2\text{O}_3$  and  $\text{Co}_{95}\text{P}_5$  (b) powders. The shiny areas in (a) and (b) correspond to Co, whereas the darker ones to  $\text{Al}_2\text{O}_3$ . (c) The hysteresis loops of the bulk materials prepared from the composite particles (1) and the mixture of powders (2).

the contribution of the ferromagnetic phase. The second term represents the contribution of the superparamagnetic ultra-thin ferromagnetic clusters not coupled by exchange. The fitting results indicate that the  $V_{sp}$  significantly increases with milling duration. The  $V_{sp} \sim 2\%$  for initial particles and  $V_{sp} \sim 12\%$  for  $\text{Al}_2\text{O}_3/\text{Co(P)}$  particles milled during 75 min. This is consistent with the decrease of magnetization value of  $\text{Al}_2\text{O}_3/\text{Co(P)}$  powders during the milling process. The magnetization is 54 emu/g for initial  $\text{Al}_2\text{O}_3/\text{Co(P)}$  particles and it is equal to 22 emu/g for  $\text{Al}_2\text{O}_3/\text{Co(P)}$  particles milled during 75 min. Such a decrease of magnetization value has not been observed for  $\text{Co}/20\text{mass}\%-\text{Al}_2\text{O}_3$  and  $\text{Co}/50\text{mass}\%-\text{Al}_2\text{O}_3$  for milling duration up to 20 h.<sup>8</sup> The decrease of magnetization value was observed for  $\text{Al}_2\text{O}_3/42\text{ mass}\%-\text{Co(P)}$  powders during the milling process.<sup>7</sup> It is likely that the magnetization decreases depends on the milling energy (i.e., mechanically driven transformation).

Information on the local anisotropy field and the grain size in the systems of exchange-coupled grains can be obtained from investigation of approach magnetization to saturation law.<sup>16–18</sup> Approach to magnetic saturation curves in the applied fields up to  $7 \div 50$  kOe for the all nanocomposites follow Akulov's law:  $M(H) = M_0(1 - aH_a^2/H^2)$ , where  $H_a = 2K/M_s$  is the local magnetic anisotropy field and  $a$  is the symmetry coefficient. This allows us to calculate the local magnetic anisotropy field  $H_a$ . Despite the increase in microstrain, the  $H_a$  value decreases from 8.4 kOe to 3.8 kOe with an increase in milling duration up to 75 min. Such  $H_a$  behavior reflects the transformation of highly anisotropic Co hcp phase to Co fcc phase during the milling process.

Figure 4 shows SEM images of bulk cermet produced by dynamic compaction of the composite  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  powder (a)—sample I and mixture of  $\text{Al}_2\text{O}_3$  and  $\text{Co}_{95}\text{P}_5$  powders (b)—sample II. It can be noticed that the sample II is characterized by larger inhomogeneous microstructure. Such layered microstructure of sample II causes an increase of the coercivity values up to 400 Oe. The coercivity  $H_c$  of sample I is 200 Oe (Fig. 4(c)), i.e., the  $H_c$  value basically stays constant during the compacting process. It is found that the local

magnetic anisotropy field  $H_a$  remains unchanged after the compaction process for sample II. In the case of composite particles compaction the grain size increases and it leads to a decrease of  $H_a$  value to 3 kOe.

#### IV. CONCLUSIONS

In summary, the structural and magnetic properties of  $\text{Al}_2\text{O}_3/\text{Co}_{95}\text{P}_5$  composite powders have been investigated. The ball milled composite powders consist of a mixture of hcp and fcc phases. It is found that the volume fraction of superparamagnetic particles significantly increases in the course of the milling process up to 12 vol. %. Reduction in grain size of Co leads to a decrease of the saturation magnetization. In spite of having somewhat lower value of magnetization, composite particles passed through the 45 min ball milling process possess a high enough value of ductility necessary to produce a bulk material with the best combination of magnetic and mechanical properties. In other words, we sacrifice magnetization value for better mechanical properties such as improvement in bending strength and fracture toughness. Also, the ball milling causes the  $H_a$  value to significantly decrease. Such  $H_a$  behavior reflects the transformation of the highly anisotropic Co hcp phase to the Co fcc phase during the milling process. Magnetic measurements show that the coercivity remains unchanged after dynamic compaction. Using the composite particles as starting powder for the compacting process allows us to produce a more homogeneous bulk sample than in the case of the mixture of Co and  $\text{Al}_2\text{O}_3$  powders.

#### ACKNOWLEDGMENTS

This work was partially supported by Russian Foundation of Basic Research 13-03-00476-a.

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