## MAGNETOELECTRIC PROPERTIES OF NdSc<sub>3</sub>(BO<sub>3</sub>)<sub>4</sub>

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The great interest shown in recent decades to the crystals of trigonal rare-earth borates  $RM_3(BO_3)_4$  was determined by their interesting multiferroic properties. These crystals have a noncentrosymmetric trigonal structure isometric to the huntite natural material  $CaMg_3(CO_3)_4$  that crystallizes in the space group *R*32 of the trigonal system [1]. At first, some of the ferroborates below the antiferromagnetic ordering temperature  $T_N \approx (30-40)$  K is observed spontaneous and/or induced by external magnetic field electric polarization, up to 300  $\mu$ C/m<sup>2</sup> [2].

Recently, in the paramagnetic trigonal rare-earth aluminoborates (Ho,Tm,Er)Al<sub>3</sub>(BO<sub>3</sub>)<sub>4</sub> [3], induced magnetoelectric polarization has also been observed. It can reach values exceeding those previously measured in ferroborates, for example, the magnetoelectric polarization of HoAl<sub>3</sub>(BO<sub>3</sub>)<sub>4</sub> achieves 4500  $\mu$ K/m<sup>2</sup> [4]. A sufficiently large induced magnetoelectric polarization was also observed in the holmium galloborate HoGa<sub>3</sub>(BO<sub>3</sub>)<sub>4</sub>, up to 1000  $\mu$ C/m<sup>2</sup> [5].

In this work, we report the data on the magnetoelectric properties of the  $NdSc_3(BO_3)_4$  single crystal. In this compound, the system of small cations is presented by  $Sc^{3+}$  ions with a larger ionic radius as compared with  $Al^{3+}$  and different electronic structure.  $NdSc_3(BO_3)_4$  single crystals were grown from bismuth trimolybdate-based fluxes. The crystal growth techniques used were described in detail in [**5**].

The magnetic and magnetoelectric properties were investigated on a PPMS-9 facility in the temperature range 3–300 K and magnetic fields up to 90 kOe.

The value of the magnetoelectric effect in  $NdSc_3(BO_3)_4$  depends not only on the size an electronic structure of  $Sc^{3+}$  ions, but also on the conditions of heat treatment of the grown crystals.



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