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## ПОВЕРХНОСТНАЯ МОДИФИКАЦИЯ НАНОЧАСТИЦ $\text{Fe}_3\text{O}_4@\text{SiO}_2$

**Петров Д.А.**

к.ф.-м.н., научный сотрудник лаборатории физики магнитных явлений  
Институт физики им. Л.В. Киренского Сибирского отделения Российской академии наук -  
обособленное подразделение ФИЦ КНЦ СО РАН

**Иванова О.С.**

к.ф.-м.н., научный сотрудник лаборатории физики магнитных явлений  
Институт физики им. Л.В. Киренского Сибирского отделения Российской академии наук -  
обособленное подразделение ФИЦ КНЦ СО РАН

Сибирский Федеральный Университет, Красноярск, 660041, Россия

**Соколов А.Э.**

к.ф.-м.н., старший научный сотрудник лаборатории физики магнитных явлений  
Институт физики им. Л.В. Киренского Сибирского отделения Российской академии наук -  
обособленное подразделение ФИЦ КНЦ СО РАН

Сибирский Федеральный Университет, Красноярск, 660041, Россия

**Эдельман И.С.**

д.ф.-м.н., профессор, главный научный сотрудник лаборатории физики магнитных явлений  
Институт физики им. Л.В. Киренского Сибирского отделения Российской академии наук -  
обособленное подразделение ФИЦ КНЦ СО РАН

**Ying Zhen Chen**

Master degree student

Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Спиваков А.А.**

Post doctor at research

Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Yaw-Teng Tseng**

Associated Professor

Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Chun-Rong Lin**

Professor

Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Аннотация.** Работа направлена на разработку технологии amino-функциональных наночастиц  $\text{Fe}_3\text{O}_4@\text{SiO}_2$ . Изучены морфология и особенности магнитных и магнитооптических свойств полученных гибридных наночастиц. Рассмотрены некоторые примеры применения таких гибридных наноструктур.

## SURFACE MODIFICATION of $\text{Fe}_3\text{O}_4@\text{SiO}_2$ NANOPARTICLES

**Petrov D.A.**

PhD, researcher of laboratory of physics of magnetic phenomena

Kirensky Institute of Physics, Federal Research Center KSC SB RAS

**Ivanova O.S.**

PhD, researcher of laboratory of physics of magnetic phenomena  
Kirensky Institute of Physics, Federal Research Center KSC SB RAS  
Siberian Federal University, Krasnoyarsk, 660041, Russia

**Sokolov A.E.**

PhD, senior researcher of laboratory of physics of magnetic phenomena  
Kirensky Institute of Physics, Federal Research Center KSC SB RAS  
Siberian Federal University, Krasnoyarsk, 660041, Russia

**Gerasimova M.A.**

Senior Lecturer of the Department of General Physics  
Siberian Federal University, Krasnoyarsk, 660041, Russia

**Edelman I.S.**

Dr.Sc., Professor  
Kirensky Institute of Physics, Federal Research Center KSC SB RAS

**Ying Zhen Chen**

Master degree student  
Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Aleksandr A. Spivakov**

Post doctor at research  
Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Yaw-Teng Tseng**

Associated Professor  
Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Chun-Rong Lin**

Professor  
Department of Applied Physics, National Pingtung University, Pingtung, Taiwan, R.O.C.

**Annotation.** *The work is aimed to the development of the technology of the amino-functionalized  $Fe_3O_4@SiO_2$  nanoparticles. The morphology and features of the magnetic and magneto-optical properties of the obtained hybrid nanoparticles are studied. Some examples of applications of such hybrid nanostructures are considered.*

**Keywords:**  *$Fe_3O_4$  nanoparticles, silica coated nanoparticles, water pollutions adsorption, magnetic properties*

Silica is frequently used as a support-material in core-shell structures, it not only helps nanoparticles (NPs) to become stable at a certain condition, but also provides the opportunity to be easily modified with other functional groups and, additionally, it has high thermal and chemical stability and, most importantly, shows environmental compatibility. Furthermore, the silica coated magnetic NPs can be dispersed in water without adding other surfactants due to the negative charges on the silica shell.

The fields of application of  $Fe_3O_4@SiO_2$  NPs functionalized with different surfactants, more frequently with amines, are varied. A number of authors have demonstrated their effective applications in biology and medicine [1, 2], in catalysis [3], and, especially, as effective adsorbents for removal of pollutants from wastewaters [4, 5, 6, 7, 8] A new type of magnetic fluorescent

nanocomposite ( $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2/\text{CQDs}$ ) was prepared by bonding of carbon quantum dots (CQDs) with  $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2$  nanocomposites through amine-carbonyl interactions and used as a fluorescent probe to detect  $\text{Cu}^{2+}$  [9]. Since this line of research is rather new, on the one hand, and the properties and application possibilities of nanoparticles strongly depend on the details of their synthesis, the search for optimal synthesis conditions and the study of the properties of functionalized particles by various methods can be considered as an urgent task.

In this study, we prepared  $\text{Fe}_3\text{O}_4$  NPs and a series of nanocomposites based on them:  $\text{Fe}_3\text{O}_4@\text{SiO}_2$ ,  $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2$ ,  $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2\text{-EY}$ , conducted a comparative study of their properties, and estimated possibilities of their application as adsorbents of water pollutions.

Magnetite  $\text{Fe}_3\text{O}_4$  NPs were synthesized by co-precipitation method using hydrolysis of iron sulfate. After that, the particles were selenized ( $\text{Fe}_3\text{O}_4@\text{SiO}_2$ ) and derivatized ( $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2$ ) for covalent attachment of Eosin Y ( $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2\text{-EY}$ ). The Eosin Y is a brominated fluorescein dye derivative used as an acidic red stain for cell cytoplasm. The synthesized samples were examined with X-ray diffraction (XRD), transmission electron microscope (TEM), Fourier-transform infrared spectroscopy (FTIR), vibrating sample magnetometer (VSM), magnetic circular dichroism spectroscopy (MCD), and fluorescence measurement.

XRD patterns revealed that all the parent NPs and magnetic core of all composite NPs were of spinel ferrite crystal structure with parameters of the most intense peaks corresponding to the  $\text{Fe}_3\text{O}_4$  phase (PDF Card # 04-005-4319). The TEM images (Fig. 1a) show that the nanoparticles synthesized by co-precipitation method are well-dispersed and have quadrangular shape. Most of the particles were 30 nm in size, and there was also a small fraction of smaller particles 5-10 nm in size. After selenized, the  $\text{SiO}_2$  covered NPs can be observed (Fig. 1b).

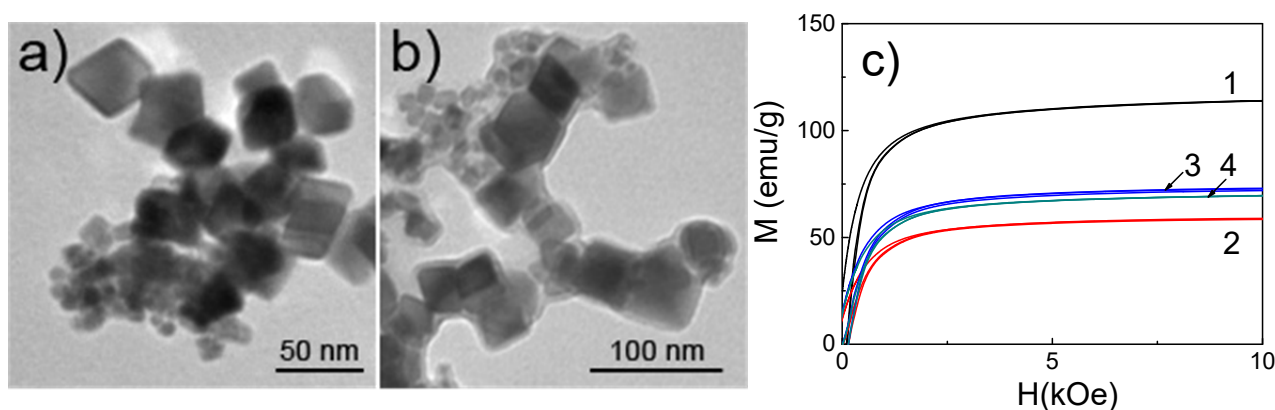


Figure 1. TEM images of  $\text{Fe}_3\text{O}_4$  NPs (a) and  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  NPs (b), room temperature magnetization curves for magnetite nanoparticles:  $\text{Fe}_3\text{O}_4$ ,  $\text{Fe}_3\text{O}_4@\text{SiO}_2$ ,  $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2$ , and  $\text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2\text{-EY}$ , curves 1-4, correspondingly (c).

The FTIR spectra show the appearance of new bands upon the transition  $\text{Fe}_3\text{O}_4 \rightarrow \text{Fe}_3\text{O}_4@\text{SiO}_2 \rightarrow \text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2 \rightarrow \text{Fe}_3\text{O}_4@\text{SiO}_2\text{-NH}_2\text{-EY}$  evidencing on chemical bonds Si-O-Si, H-O-H,  $\text{NH}_2$ . Magnetic measurements (Fig. 1 c) show the saturation magnetization decrease of NPs coated with a  $\text{SiO}_2$  shell. However, functionalization of silanized NPs with amino groups leads to some increase of their magnetization value. We ascribe this variety to the surface modification of NPs. Fluorescence measurement pumped at 2.76 eV (450 nm) was measured at room temperature.

MCD spectrum of  $\text{Fe}_3\text{O}_4$  NPs is characteristic for this material (Fig. 2). Silanization of the samples led to a red shift in the MCD spectra. The difference in the MCD spectra of the silanized and initial samples indirectly confirms the formation of a  $\text{SiO}_2$  shell around the  $\text{Fe}_3\text{O}_4$  NPs. The functionalization of the silanized sample did not lead to a significant change in the spectra.

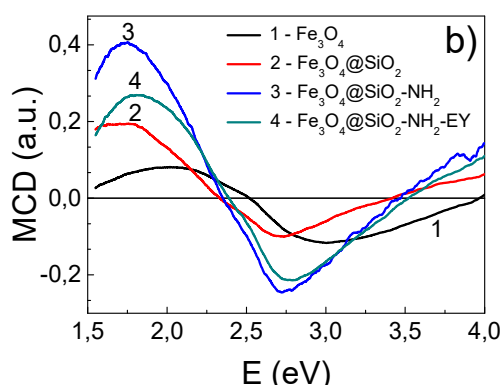


Figure 2. MCD spectra for magnetic NPs at room temperature.

So, the results described above allow concluding that our experimental process is successful and stable:  $\text{Fe}_3\text{O}_4@\text{SiO}_2$  magnetic nanoparticles with core-shell structure were prepared and functionalized for adsorption of cationic and anionic dyes and medical imaging applications.

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